Some 4-Thio-DL-tetrofuranose Derivatives and Other Oxidation Products of Thiolane-3,4-diols

By JOAN E. McCORMICK and R. S. McElhinney*

(Laboratories of the Medical Research Council of Ireland, Trinity College, Dublin 2, Ireland)

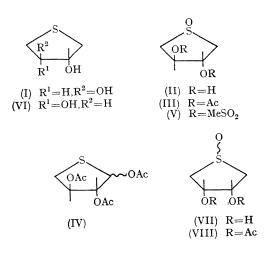
ALTHOUGH many derivatives of pentoses and hexoses in which sulphur replaces the ring oxygen atom are now known,¹ and analogues of tetroses with nitrogen as the ring hetero-atom have been prepared,² 4-thiotetrofuranose derivatives have not yet been described. We have now prepare1 such compounds from the thio-derivatives of anhydro-polyhydric alcohols; this method represents a new approach to thiofuranoses in general.

Oxidation of the DL-trans-diol (I)³ with one equivalent of hydrogen peroxide in acetic acid gave the DL-sulphoxide (II)

(96% yield; m.p. $160\cdot5-161\cdot5^{\circ}$). Two equivalents of acetic anhydride in pyridine transformed this into the diacetate (III), m.p. $82-83\cdot5^{\circ}$. When this ester (5 mmole) was treated with acetic anhydride (5.5 mmole) in boiling benzene (1 ml.),⁴ Pummerer rearrangement led to a syrupy mixture of anomeric triacetates (IV) from which one isomer, 1,2,3-tri-O-acetyl-4-thio-DL-threofuranose, m.p. $84\cdot5-85\cdot5^{\circ}$, crystallised. Hydrogen peroxide in acetic acid converted this into the corresponding sulphone, m.p. $124-127^{\circ}$.

Treatment of the diol (II) with methanesulphonyl chloride and pyridine⁵ caused decomposition, but the dimethanesulphonate (V), m.p. 152-152.5°, was readily obtained from the reaction of its sulphide counterpart with peroxide by the method of Procházka.6

In the case of the cis-diol (VI)⁶ (dimethanesulphonate, m.p. 111-113°; N-tosyl sulphilimine, m.p. 169-170°), the same reaction sequence as outlined for the trans-diol



was complicated by the formation of two isomeric sulphoxides in the initial oxidation. Attempts to separate these with mercuric chloride⁷ were not promising, so a product, m.p. 100—103 \cdot 5°, consisting mainly of one isomer (VII) was acetylated. The resulting diester (VIII), m.p. 64— 67.5° , when subjected to the Pummerer rearrangement again gave a syrupy mixture from which a crystalline product, 1,2,3-tri-O-acetyl-4-thio-DL-erythrofuranose, m.p. 103-104°, was readily isolated by trituration with water. The derived sulphone had m.p. 168-170°.

Oxidation of the diols (I) and (VI) with two equivalents of hydrogen peroxide in acetic acid readily gave the known sulphones.⁸ One equivalent of sodium periodate converted the cis-diol (VI) into the sulphide dialdehyde in 10 min. at 2° , whereas two equivalents yielded the sulphoxide dialdehyde in 5 hr. The latter was also obtained in high yield from the mixture of sulphoxides (VII) in 10 min. One equivalent of periodate reacted with cis-sulpholane-3,4-diol to give the sulphone dialdehyde. The three dialdehydes were isolated⁹ from these reactions in 80-85% yield as the corresponding bis-(4-methylthiosemicarbazones), m.p. 185-186°, 189-190°, and 192.5° respectively.

We thank Dr. V. C. Barry for his interest and encouragement, and May and Baker Ltd., Dagenham, Essex, and the Irish Cancer Society for financial support.

(Received, December 9th, 1968; Com. 1681.)

- ¹ H. Paulsen, Angew. Chem. Internat. Edn., 1966, 5, 508; J. M. Cox and L. N. Owen, J. Chem. Soc. (C), 1967, 1121. ² W. A. Szarek and J. K. N. Jones, Canad. J. Chem., 1965, 43, 2345.
- ³G. W. Kilmer, M. D. Armstrong, G. B. Brown, and V. du Vigneaud, J. Biol. Chem., 1942, 145, 495.
- ⁴ L. Horner and P. Kaiser, Annalen, 1959, 626, 19.
- ⁵ D. N. Jones, M. J. Green, and M. A. Saeed, Chem. Comm., 1967, 674.
- ⁶ M. Procházka, Coll. Czech. Chem. Comm., 1965, 30, 1158.
- 7 H. B. Henbest and S. A. Khan, Proc. Chem. Soc., 1964, 56; C. R. Johnson and D. McCants, J. Amer. Chem. Soc., 1965, 87, 1109, 5404.
- ⁸ M. Procházka and V. Horák, Coll. Czech. Chem. Comm., 1959, 24, 1509.
- V. C. Barry, J. E. McCormick, and R. S. McElhinney, Carbohydrate Res., 1968, 7, 299.